

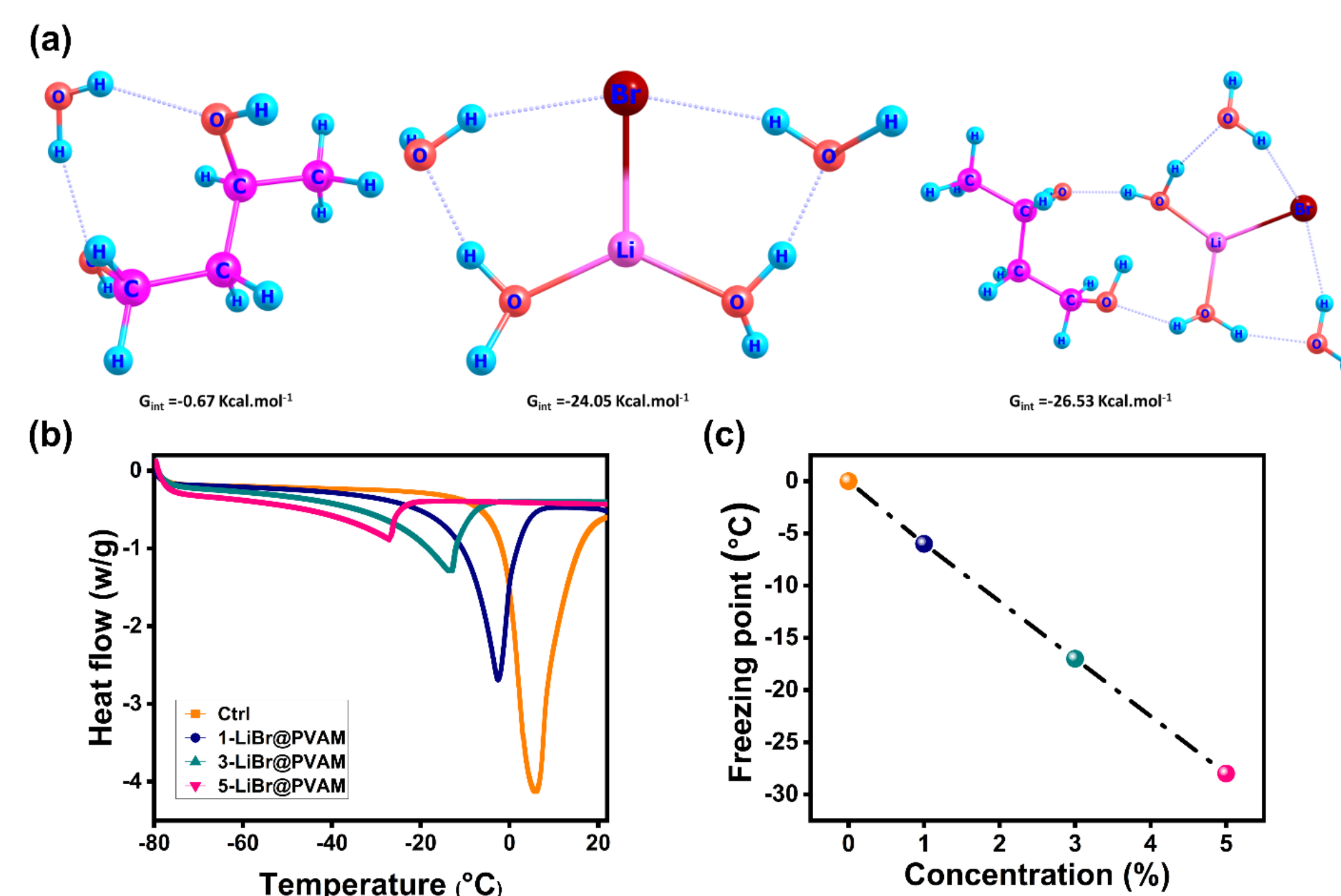
Novel Non-Flammable and Self-Regenerative High-Performance Hydrogel Electrolyte with Anti-Freeze Properties and Intrinsic Redox Activity for Energy Storage Applications

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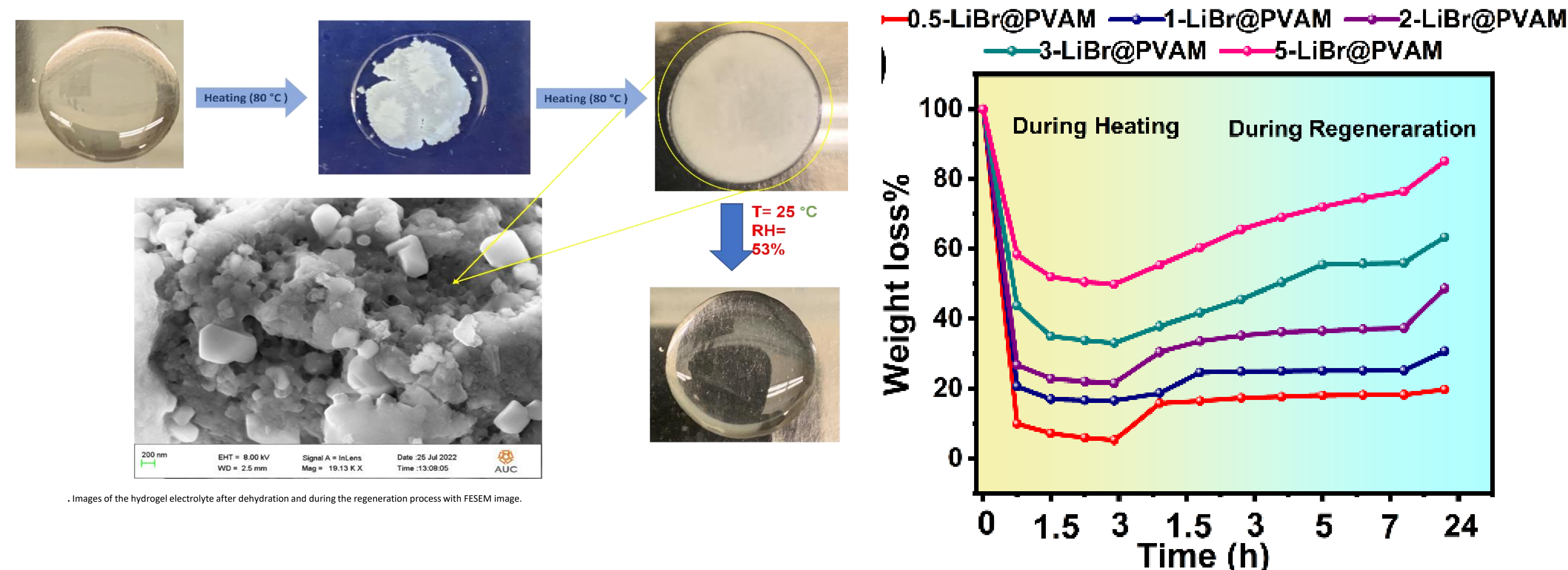
Introduction

- Hydrogel electrolytes are essential components of a plethora of functional devices due to their flexibility and high electronic and ionic conductivity.
- Hydrogel Electrolytes suffer from poor water retention (dehydration) during operation.
- The overall performance of the hydrogels-based devices is severely declined as a result of conductivity fading of the hydrogel with poor self-regeneration.
- To this end, the rational tailoring of hydrogel electrolytes with high conductivity, self-regeneration, non-flammability, anti-freezing ability, stability, and intrinsic redox activity is necessary to enable the fabrication of highly durable devices.
- Herein, we demonstrate the design and synthesis of highly ionic conductive LiBr@PVA-based electrolytes. Upon the use of the synthesized hydrogel electrolytes in supercapacitor devices, they revealed intrinsic redox activity with outstanding water retention capability and self-regeneration characteristics.

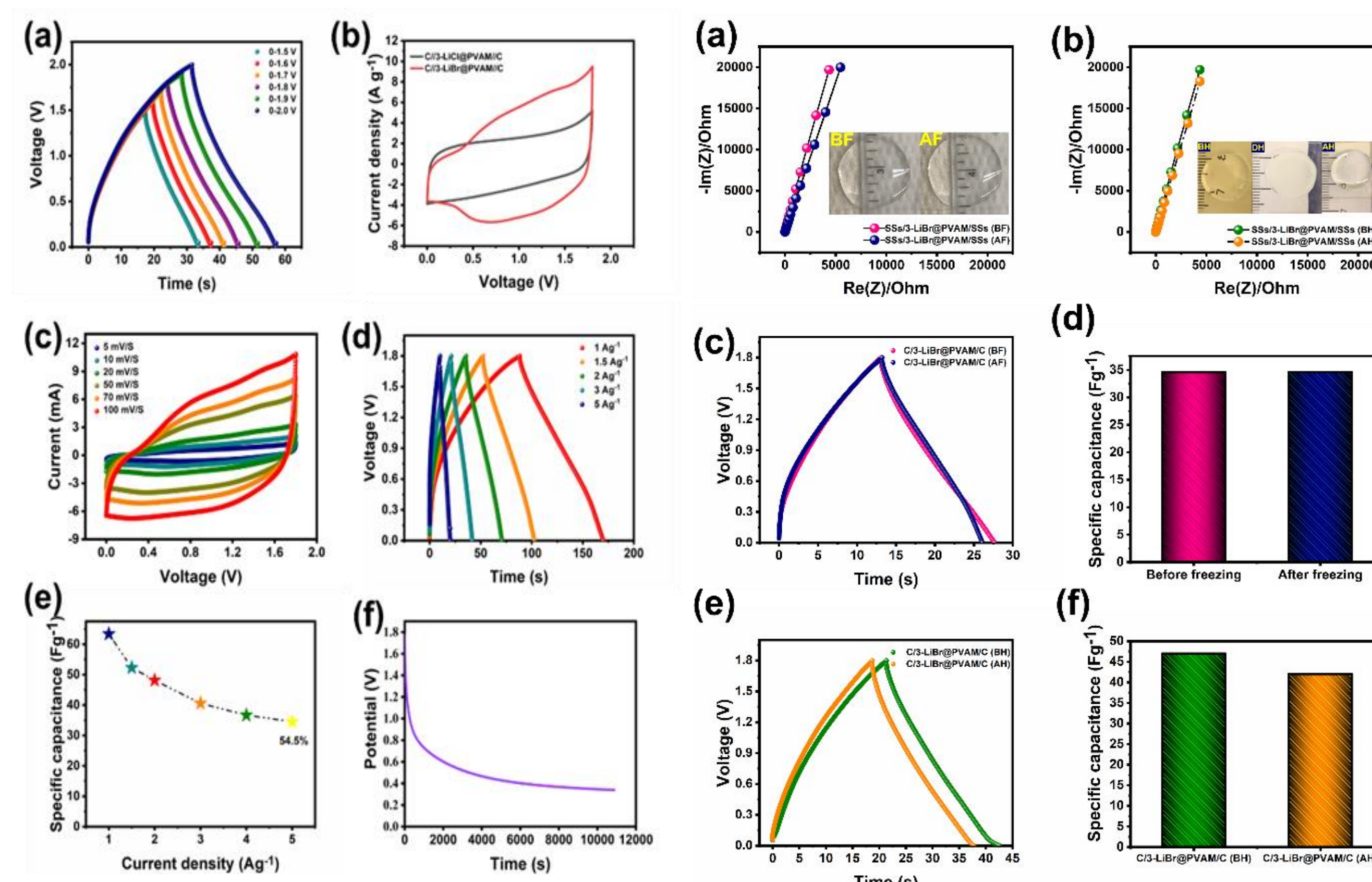
Water Retention Measurements



Water Regeneration Measurements

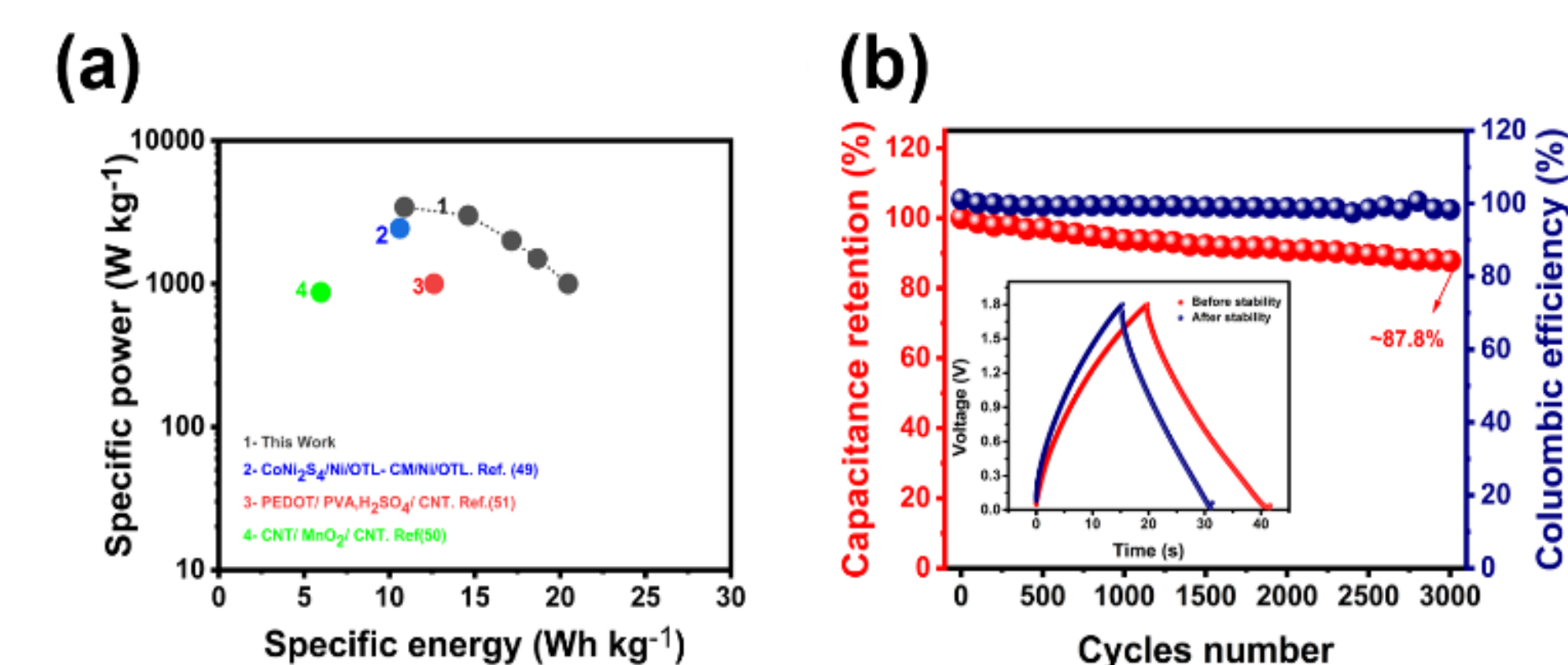


Highlighted Electrochemical Results



Electrochemical Results

- The hydrogel electrolyte showed higher ionic conductivity than those reported in the literature.
- Upon testing the hydrogel as an electrolyte in supercapacitor devices, the fabricated C//3-LiBr@PVAM//C device showed high operating potential window of 1.8 V and took more than 10,800 s to drop from 1.8 V to 0.3 V during the self-discharge testing.
- In addition, the intrinsic redox properties of the electrolyte caused an increase in the specific capacitance, reaching 63.3 F/g.
- The fabricated device exhibited high specific capacitance compared to similar systems reported in the literature. The device can deliver high energy density of ~ 20.5 Wh/kg with a power density of 3430 W/kg.



Conclusion

We demonstrate a facile synthesis approach of novel hydrogel electrolytes based on LiBr@PVA and their utilization in energy storage devices. The electrolyte can regenerate more than 70% of its water content via absorption of water from the surrounding environment within 24 hours. The DFT calculations unraveled the reason behind the high-water retention ability of the fabricated hydrogel. Also, the DSC analysis revealed the anti-freezing properties of the hydrogel electrolyte, retaining its liquid state at -30 °C. Moreover, the electrolyte is considered a completely non-flammable composite for use as a safe electrolyte for solid state devices.

References

- Qin W, Zhou N, Wu C, Xie M, Sun H, Guo Y, et al. Mini-Review on the Redox Additives in Aqueous Electrolyte for High Performance Supercapacitors. ACS Omega. 2020;5(8):3801–8.
- Chen M, Zhang Y, Xing G, Chou S-L, Tang Y. Electrochemical energy storage devices working in extreme conditions. Energy Environ Sci. 2021;14(6):3323–51.
- Zhou Y, Qi H, Yang J, Bo Z, Huang F, Islam MS, et al. Two-birds-one-stone: multifunctional supercapacitors beyond traditional energy storage. Energy Environ Sci. 2021;14(4):1854–96.
- Hashemi M, Rahmanifar MS, El-Kady MF, Noori A, Mousavi MF, Kaner RB. The use of an electrocatalytic redox electrolyte for pushing the energy density boundary of a flexible polyaniline electrode to a new limit. Nano Energy. 2018;44:489–98.
- Shao Y, El-Kady MF, Wang LJ, Zhang Q, Li Y, Wang H, et al. Graphene-based materials for flexible supercapacitors. Chem Soc Rev. 2015;44(11):3639–65.
- Simon P, Gogotsi Y, Dunn B. Where do batteries end and supercapacitors begin? Science 2014;343(6176):1210–1.
- Zhang L, Yang S, Chang J, Zhao D, Wang J, Yang C, et al. A Review of Redox Electrolytes for Supercapacitors. Front Chem. 2020;8:1–7.
- El Sharkawy HM, Dhmees AS, Tamman AR, El Sabagh SM, Aboushabha RM, Allam NK. N-doped carbon quantum dots boost the electrochemical supercapacitive performance and cyclic stability of MoS₂. J Energy Storage. 2020;27:101078.

Why This Work ?

- Solid State Electrolyte
- Water Retention
- Safety Considerations
- Redox Activity
- Anti freezing Properties
- Patent

This Work Achievements

- Published JMCA, RSC IF=15
- Filed as USUS Provisional Patent no. 1633/15 PROV
- 6 Citations in just 6 months